

Appendix A

Descriptors

Descriptors for Torsion Angles and ring substituents

Table A.0.1: Descriptors for Torsion Angles

Torsion Angle Range, ($^{\circ}$)	Full Descriptor	Short Descriptor
0 to 30	+ Syn-Periplanar	+ sp
30 to 90	+ Syn-Clinal	+ sc
90 to 150	+ Anti-Clinal	+ ac
150 to 180	+ Anti-Periplanar	+ ap
0 to -30	- Syn-Periplanar	- sp
-30 to -90	- Syn-Clinal	- sc
-90 to -150	- Anti-Clinal	- ac
-150 to -180	- Anti-Periplanar	- ap

Table A.0.2: Descriptors for Ring Substituents

Angle Range of Substitution	Full Descriptor	Short Descriptor
0 to 30	Axial	ax
30 to 60	Bisectional	bi
60 to 90	Equatorial	eq



Colophon

This manuscript was typeset by the author on a PC running Debian GNU/Linux (Sarge).

The details of software packages used are as follows:

Text editing	Vim
	http://vim.sourceforge.net
Type setting	L ^A T _E X
Coverpage	front-Gdis
	http://freshmeat.net/projects/gdis
	back-XdisplayF Z. Otwinowski et al [†] .
Schematic diagrams	Chemtool
	http://freshmeat.net/projects/chemtool
Molecular diagrams	PLATON A. L. Spek ‡
Graph	Gnuplot
	http://gnuplot.sourceforge.net

[†]Z. Otwinowski and W. Minor, "Processing of X-ray Diffraction Data Collected in Oscillation Mode", *Methods in Enzymology, Volume 276: Macromolecular Crystallography, part A*, p.307-326, 1997, C.W. Carter, Jr. & R.M. Sweet, Eds., Academic Press.

[‡]A. L. Spek, *J. Appl. Cryst.*, **36**, 7-13, 2003.

The body type is 10 point Computer Modern Roman. Chapter and section titles are in various sizes of Adobe Helvetica-Narrow Bold. The final output was converted to a PDF file, which was then printed at ADS India, Mysore.



Publications

Pertaining to the research work

1. Crystal and Molecular Structure of 3,5-Dicarbethoxy-2, 6-dimethyl-4-(4-methoxy)phenyl-1,4-dihydropyridine: Lakshmi Srinivasan, Sridhar M. Anandalwar, Javaregowda S. Prasad, Dinesh Manvar, Alpesh Parecha, Gautam Patel, Anamik Shah, *J. Anal. Sci.*, **21**, 93–94, 2005.
2. Synthesis and Molecular Structure Analysis of Venlafaxine Intermediate and its Analog, Kavitha C. V., Lakshmi Srinivasan, Basappa, Rangappa Kanchugarakoppal Subbegowda, Mantelingu K., Sridhar M. Anandalwar, Shashidhara Prasad J.: *Jour. of Chem. Cryst.*, (accepted), Manuscript No. JOCC207R1.

Communicated/to be communicated

1. Crystal and Molecular Structure of 2,6-dimethyl-3, 5-di-N- (2'-chlorophenyl) carbamoyl-4-(4''-hydroxy phenyl)-1,4-dihydropyridine: Lakshmi Srinivasan, Sridhar M. Anandalwar, J. Shashidhara Prasad, Dinesh Manvar, Rajesh Loria, Gautam Patel, Anamik Shah. (*Communicated to J. Anal. Sci.*), 2005.
2. Crystal structure of 2,6-dimethyl-3,5-di-N-phenylcarbamoyl-4(3'- nitro phenyl)-1,4-dihydropyridine:Lakshmi Srinivasan, Sridhar M. Anandalwar, J. Shashidhara Prasad, Chintan Dholakia, Dinesh Manvar, Vishal Narodia, Anamik Shah.

3. Crystal structure of 2-amino-4(3'-Chloro Phenyl)-3-ethoxycarbonyl-4H-pyrano-[3,2-c]-chromene-5-one: Lakshmi Srinivasan Sridhar M. Anandalwar, Shashidhara Prasad J., Dinesh Manvar, Alpesh Parecha, Gautam Patel, Jignesh Patel, Anamik Shah.
4. Crystal structure confirmation and comparison of pyrano-[3,2-c]-chromene derivatives: Lakshmi S., Sridhar M. A., Shashidhara Prasad J., Dinesh Manvar, Alpesh Parecha, Jignesh Patel, Anamik Shah.

Others

1. Structural Analysis of 4, 4'-Methoxy Bis Hydrazone: S. Lakshmi, M.A. Sridhar, J. Shashidhara Prasad, J. Indira, Prakash P. Karat, B. Veerendra, B. Shivarama Holla. *Mol. Cryst. Liq. Cryst.*, **381**, pp 59-68, 2002.
 2. Synthesis and Structural Analysis of Tetrathioureacopper(I) Chloride: Lakshmi S., Sridhar M.A., Shashidhara Prasad J., Srinivasan V., Kandhaswamy M. A., Dhandapani M., *J. Anal. Sci.*, **19**, 19-20, 2003.
 3. Synthesis and Structural Analysis of Tetraethylammonium Trichloro-cadmate: S. Lakshmi, M. A. Sridhar, J. Shashidhara Prasad, G. Amirthaganesan, M. A. Kandhaswamy, V. Srinivasan, *J. Anal. Sci.*, **20**, 57-58, 2004.
 4. Growth, characterization and crystal structure analysis of 1-(4-chlorophenyl)-3-(4-chlorophenyl)-2-propen-1-one: Vincent Crasta, V. Ravindrachary, S. Lakshmi, S. N. Pramod, M. A. Shridar and J. Shashidhara Prasad, *Journal of Crystal Growth*, **275**, e329-e335, 2005.
-

Crystal and Molecular Structure of 3,5-Dicarboxy-2,6-dimethyl-4-(4'-methoxy)phenyl-1,4-dihydropyridine**Lakshmi SRINIVASAN,* Sridhar M. ANANDALWAR,** Javaregowda S. PRASAD,*
Dinesh MANVAR,** Alpesh PARECHA,** Gautam PATEL,** and Anamik SHAH****

*Department of Studies in Physics, University of Mysore, Mysore 570 006, India

**Department of Chemistry, Saurashtra University, Rajkot 360 005, Gujarat, India

1,4-Dihydropyridines (DHP) are known for their action as calcium channel blockers and are used for treatment of various cardiovascular diseases. The title compound was synthesised following the classical Hantzsch synthesis. The compound crystallises in monoclinic crystal class in the space group $P2_1/n$ with cell parameters $a = 9.744(3)\text{Å}$, $b = 7.460(2)\text{Å}$, $c = 26.662(5)\text{Å}$, $\beta = 98.34(2)^\circ$, $Z = 2$. The structure adopts a flat boat conformation.

(Received November 30, 2004; Accepted April 26, 2005; Published on web , 2005)

1,4-Dihydropyridines are known for their action as calcium channel blockers and are used for treating various cardiovascular diseases. Most of the 1,4-dihydropyridine compounds possess different functional groups on the C4 phenylring.¹ 1,4-Dihydropyridine compounds are widely prescribed for the treatment of hypertension and heart deliriation. The activity is believed to arise from binding with a receptor site located in the α_1 subunit of the L-type voltage gated channels present in skeletal and cardiac muscle.² For DHP molecules, structure-activity relationship studies³ have

indicated specific conformational details which correlate with high binding efficiency.⁴

X-ray crystallographic studies of molecules of dihydropyridine have established the fact that a majority of 1,4-dihydropyridine rings have a boat type conformation with varying degrees of puckering at C4 position.⁵ Modifications to the C4 part of 1,4-dihydropyridine moiety have been carried out. The title compound was prepared by Hantzsch (Dihydro) pyridine synthesis. A schematic diagram of the molecule is shown in Fig. 1.

A mixture of 4-methoxybenzaldehyde (1.2 g, 0.01 mol) and ethylaceto acetate (3.18 g, 0.025 mol) in methanol (40 ml) was stirred and refluxed for 5 to 10 min. Then ammonia solution (4 ml) was added to it. It was stirred for 15 to 36 more h. The resulting solid product was separated out and filtered. The yield was 58%. The purity of the compound was checked by TLC (acetone:benzene = 4:6). The compound was recrystallized

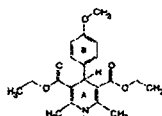


Fig. 1. Schematic diagram of the molecule.

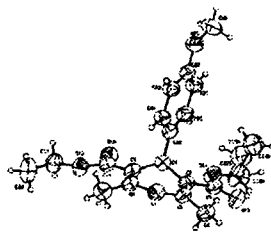


Fig. 2. ORTEP of the molecule at 30% probability.

* To whom correspondence should be addressed.
E-mail: mas@physics.uni-mysore.ac.in

Table 1. Crystal data and structure refinement table

Empirical formula	$C_{21}H_{23}NO_6$
Formula weight	359.41
$\lambda(\text{Mo K}\alpha)$	0.71073 Å
Crystal system	monoclinic
Space group	$P2_1/n$ $Z = 4$
Cell dimensions	$a = 9.744(3)\text{Å}$ $b = 7.460(2)\text{Å}$ $c = 26.662(5)\text{Å}$ $\beta = 98.34(2)^\circ$
Volume	1917.6(9) Å ³
D_x	1.245 Mg/m ³
D_{calc}	28.28
F	0.0975
$(\Delta\rho)_{\text{max}}$	0.000
$(\Delta\rho)_{\text{min}}$	0.23 e Å ⁻³
$(\Delta\rho)_{\text{res}}$	-0.265 e Å ⁻³
Refinement method	full-matrix least-squares on F^2
Measurement	DIPLABO kappa
Program system	Denzo
Structure determination	SHELXS-97 [†]
Refinement	full-matrix SHELXL-97 [†]

Table 2 Atomic coordinates and equivalent thermal parameters of the non-hydrogen atoms

Atom	x	y	z	U _{eq}
N1	0.1097(3)	-0.1435(4)	-0.2388(1)	0.0962(9)
C2	0.3471(4)	-0.1181(6)	-0.0736(2)	0.0860(9)
C3	0.3196(5)	0.0468(7)	-0.0527(3)	0.0834(9)
C4	0.2289(5)	0.1694(7)	-0.0991(3)	0.0929(9)
C5	0.3158(4)	0.1823(6)	-0.0301(2)	0.0696(9)
C6	0.2383(4)	-0.0688(5)	-0.0311(2)	0.0677(9)
C7	0.3343(4)	-0.0941(6)	0.0322(2)	0.0429(4)
C8	-0.0374(4)	-0.2844(7)	0.0653(2)	0.1097(1)
C9	-0.1001(5)	-0.2843(7)	-0.1294(2)	0.0891(1)
O10	-0.1907(3)	0.0257(5)	-0.1469(1)	0.1186(1)
O11	-0.1124(3)	0.2076(5)	-0.1237(1)	0.1000(1)
O12A	-0.2677(2)	0.3067(4)	-0.1535(9)	0.0967(6)
O12B	-0.2397(2)	0.4087(4)	-0.3017(1)	0.1274(7)
O12C	-0.2244(2)	0.2111(5)	-0.1742(2)	0.1396(9)
O12D	-0.1772(2)	0.3301(4)	-0.2290(6)	0.1192(1)
O14	0.2754(4)	0.3741(5)	-0.0046(2)	0.0812(1)
O15	0.2715(3)	0.2917(4)	0.0360(1)	0.0989(8)
O16	0.2196(3)	0.4123(5)	-0.0108(1)	0.1055(9)
O17	0.2294(5)	0.4445(6)	0.0607(2)	0.1057(1)
O18	0.1114(6)	0.3763(9)	0.1116(3)	0.1534(2)
O19	0.2198(6)	0.3098(9)	-0.1288(2)	0.0821(1)
O20	0.1815(5)	0.2004(6)	-0.1648(2)	0.0913(1)
O21	0.2549(4)	0.2171(9)	-0.2019(2)	0.0649(1)
O22	0.2074(5)	0.2277(6)	-0.2001(1)	0.0893(1)
O23	0.4006(4)	-0.1101(8)	-0.1608(2)	0.0895(1)
O24	0.2313(3)	0.1368(5)	-0.1254(2)	0.0853(1)
O25	0.1378(3)	0.2229(5)	-0.2405(1)	0.1023(9)
O26	0.4059(6)	0.2093(9)	-0.2859(2)	0.1371(2)

[U_{eq} = (1/3)(U₁₁ + U₂₂ + U₃₃)] (a, b, c)
 *Occupancy factor is 0.33(3).
 †Occupancy factor is 0.65(3).

Table 3 Selected Bond Lengths (Å) and Angles (°)

Atoms	Length	Atoms	Length
N1-C2	1.372(5)	O25-C26	1.427(5)
N1-C6	1.378(5)	O12A-C13A	1.394(4)
C2-C19	1.509(5)	O12B-C13B	1.362(5)
C5-C14	1.437(5)	C14-O16	1.217(5)
C5-O16	1.215(5)	C14-O15	1.208(5)
C8-O11	1.372(5)	O15-C17	1.452(5)
C19-C24	1.385(5)	C17-C18	1.457(7)
Angle			
C2-C1-N1	116.4(3)	O18-C8-C2	126.8(1)
C3-C2-C8	127.5(4)	O11-C8-C2	112.5(4)
N1-C2-C8	113.2(4)	C12B-O11-C9	119.9(1)
C2-C3-C4	121.0(4)	C9-O11-C12A	114.4(1)
C3-C2-C4	118.9(3)	C13A-C12A-O11	105.7(2)
C6-C3-C4	115.9(4)	O11-C12B-C13B	113.1(2)
C5-C4-C19	112.6(3)	O16-C14-O15	128.9(1)
C5-C4-C3	110.9(3)	O16-C14-C8	123.9(4)
C19-C3-C3	110.0(3)	O15-C14-C3	115.7(4)
C6-C3-C3	125.1(4)	C14-O15-C17	117.7(3)
C8-C2-C4	119.8(3)	O15-C17-C18	106.5(4)
C14-C5-C4	115.9(3)	C24-C25-C26	121.2(3)
C5-C6-N1	118.7(4)	O25-C25-C26	115.7(3)
C5-C6-C7	126.4(4)	O25-C25-C23	115.7(3)
N1-C6-C7	112.7(4)	C23-C25-C26	117.1(3)
O16-C7-C11	121.1(4)		

C2-C3-C9-O10 are -170.31(4)° and -5.95(7)° respectively. These are comparable with the torsion angles reported for the methoxy-substituted diethyl 4-phenyl-2, 6-dimethyl-1, 4-dihydropyridine-3, 5-dicarboxylate compounds.⁴

from dimethylformamide (DMF) and methanol, m. p. 116°C. [Required C (66.85%), H (6.96%), N (3.89%)] [Found C (66.82%), H (6.94%), N (3.85%)].

A 2.0 g amount of compound was taken in 15 ml of DMF. Charcoal (1.5 g) was added and the solution was heated for 2 min. The solution was filtered while hot through Whatman 42 filter paper. The solution was kept in a slightly opened stopper conical flask for 25 days. Light yellow crystals were grown by thin film evaporation technique.

A single crystal of the title compound with dimensions 0.2 × 0.3 × 0.3 mm was chosen for X-ray diffraction studies. The data were collected on a DIFLAB Image Plate system with graphite monochromated radiation (Mo K_α). Thirty six frames of data were collected in oscillation mode with an oscillation range of 5° and processed using Denzo.⁵ The reflections were merged with Scalepack.⁶ Figure 2 shows the ORTEP of the molecule at 30% probability. The crystal and experimental details are given in Table 1.

Table 2 gives the atomic coordinates and equivalent thermal parameters of the non-hydrogen atoms. The hydrogen atoms were fixed at chemically acceptable positions and were refined with isotropic temperature factors. Tables 3 and 4 give the bond lengths and angles of non-hydrogen atoms. The carbon atoms C12 and C13 of the dicarboxy group at C5 exhibit positional disorder. However, atoms of dicarboxy group at C3 do not show any disorder.

The pyridine ring (N1, C2, C3, C4, C5, C6) conforms to a flattened boat configuration with the deviation of N1 and C4 being 0.144(3)Å and 0.290(3)Å respectively indicating higher binding efficiency.⁴ The phenyl ring is planar. The torsion angle about C4-C5-C14-O15 is -168.7(3)° and that about C4-C3-C9-O11 is -11.2(3)°, showing that the dicarboxy group at C3 and C5 do not conform symmetrically about the pyridine ring. The torsion angles C6-C5-C14-O16 and

Acknowledgements

The authors are thankful to Department of Chemistry, Saarashtra University, Rajkot for providing Laboratory facility and to DST, Government of India, for financial assistance under the project SP/12/FGO/93.

References

1. M. Mahendra, B. H. Doraswamy, A. Parecha, J. Patel, A. Shah, M. A. Sridhar, and J. S. Prasad, *Anal. Sci.*, **2004**, *20*, x19.
2. T. Tamabe, H. Takeshima, A. Mikami, H. Takahashi, K. Kangawa, M. Keyama, T. Hi-rose and Numa, *Nature*, **1987**, *328*, 315.
3. D. J. Triggle, D. A. Langa, and R. A. Janis, *Med. Res.*, **1989**, Rev.9, 123.
4. S. K. Metcalf and E. M. Holt, *Acta Cryst.*, **2000**, *C56*, 1228.
5. G. Ravnyak, N. Anderson, J. Gougoutas, A. Hedberg, S. D. Kimball, M. Malley, S. Moreland, M. Pembican, and A. Pudzanowski, *J. Med. Chem.*, **1991**, *34*, 2521.
6. G. M. Sheldrick (SHELXS-97, SHELXL-97), **1997**, University of Göttingen, Germany.
7. Z. Otwinowski and W. Minor, "Macromolecular Crystallography", **1997**, 276; part A, p. 307, ed. C. M. Carter, Jr. and R. M. Sweet, Academic Press.
8. S. Mackay, C. J. Gilmore, C. Edwards, N. Stewart, and K. Shankland, *MacXus Computer Program for the Solution and Refinement of Crystal Structures*, **1999**, Bruker Nonius, The Netherlands, MacScience, Japan, and The University of Glasgow.

Date: Apr 12, 2005

To: "Rangappa Kanchugarakoppal Subbegowda" rangappaks@yahoo.com

From: "Journal of Chemical Crystallography"

jchemcryst@chemed.ces.clemson.edu

Subject: Decision on your manuscript

Dear ksr:

I am pleased to inform you that your manuscript, "Synthesis and Molecular Structure Analysis of Venlafaxine Intermediate and its Analog" has been accepted for publication in Journal of Chemical Crystallography.

You will be contacted about proofs and offprints in due course by our Manufacturing Department.

Please remember to quote the manuscript number, JOCC207R1, whenever inquiring about your manuscript.

Congratulations and best regards,

William T. Pennington

Editor-in-Chief

Journal of Chemical Crystallography