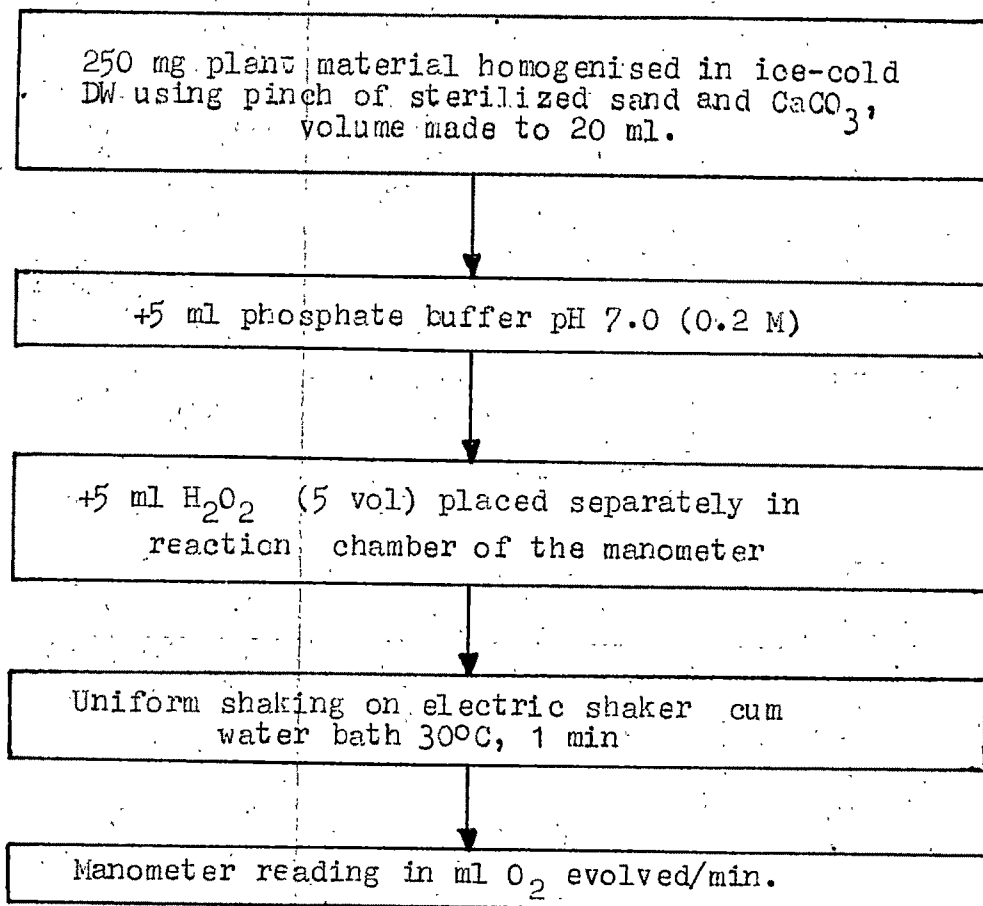


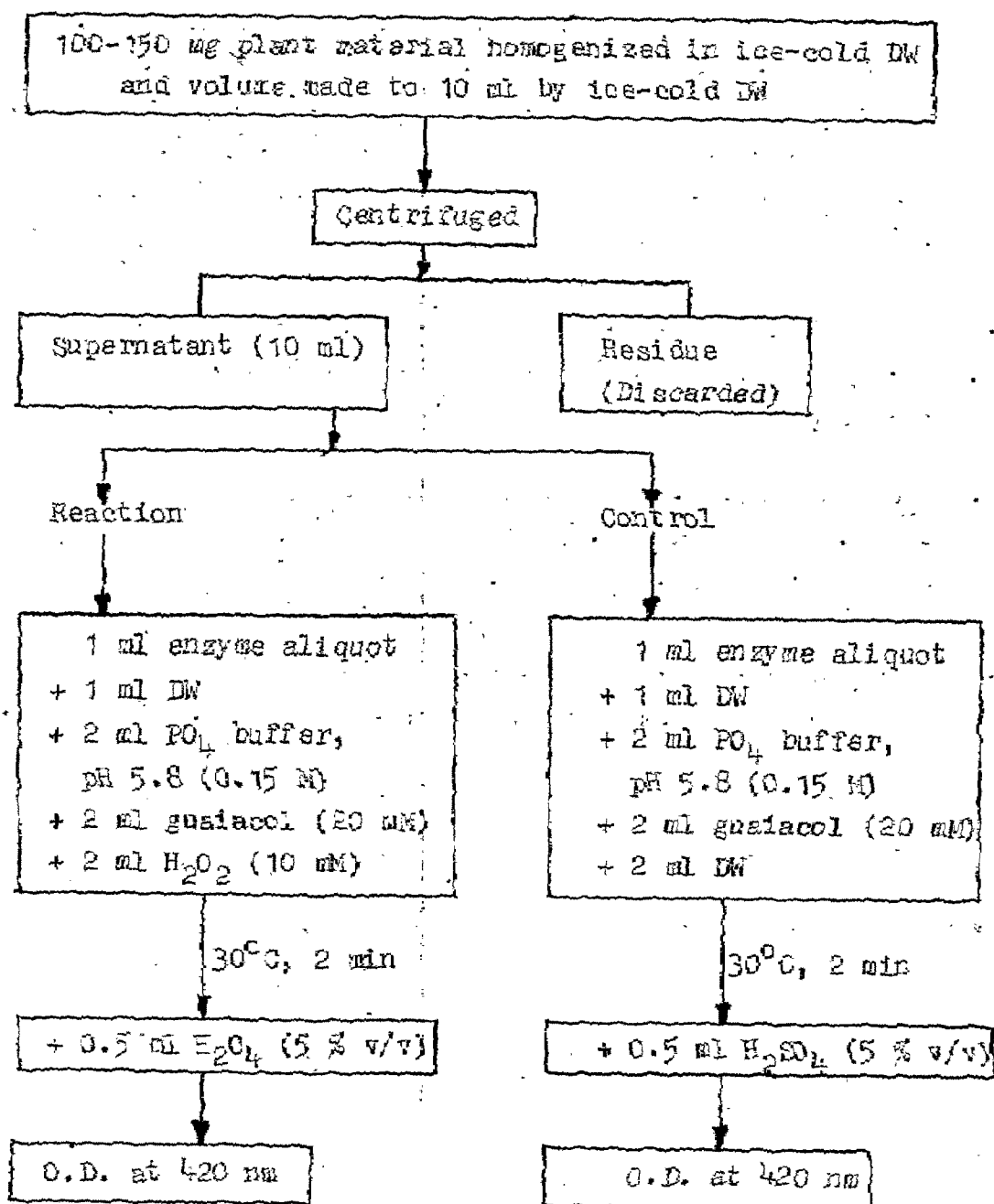
Fig. A.1 Flow Chart for Determination of Catalase Activity



## Preparation of Reagents Used

- i. Phosphate buffer pH 7.0 (0.2 M) : A solution of 31.20 g of  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  per litre and a solution of 28.39 g of  $\text{Na}_2\text{HPO}_4$  per litre (or 71.7 g of  $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  per litre) was prepared. To make 100 ml of phosphate buffer pH 7.0; 19.5 ml of  $\text{NaH}_2\text{PO}_4$  and 30.5 ml of  $\text{Na}_2\text{HPO}_4$  were solutions were added and volume made upto 100 ml by DW.
- ii.  $\text{H}_2\text{O}_2$ , 5 vol : 5 ml of 100 v/v  $\text{H}_2\text{O}_2$  (30 %  $\text{H}_2\text{O}_2$ ) were added to 95 ml DW.

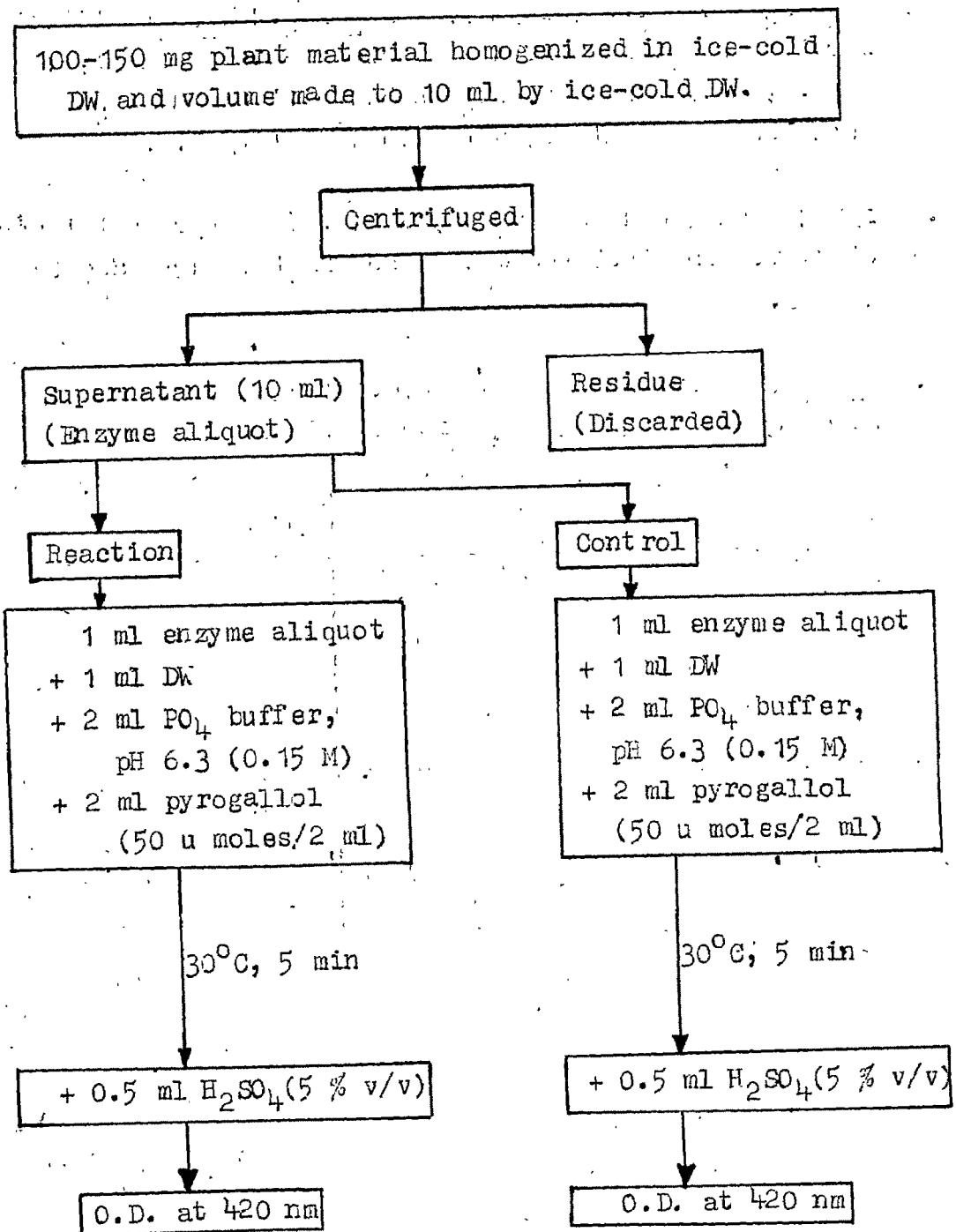
Fig. A.2 Flow Chart for Determination of Peroxidase Activity



### The Preparation of Reagents Used

- i. Phosphate buffer, pH 5.8 (0.15 M): A solution of 23.4 g of  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  per litre and a solution of 21.29 g of  $\text{Na}_2\text{HPO}_4$  per litre (or 53.77 g  $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  per litre). To make 100 ml of phosphate buffer, pH 5.8, 46.0 ml of  $\text{NaH}_2\text{PO}_4$  and 4.0 ml of  $\text{Na}_2\text{HPO}_4$  solutions were added and volume made upto 100 ml by DW.
- ii. Guaiacol (20 mM): 0.22 ml of pure guaiacol added in 99.78 ml DW, shaken vigorously and stored in a brown bottle in fridge.
- iii.  $\text{H}_2\text{O}_2$  (10 mM): 0.11 ml of 100 v/v  $\text{H}_2\text{O}_2$  (30 %  $\text{H}_2\text{O}_2$ ) added to 99.89 ml DW (or 0.57 ml of 20 v/v  $\text{H}_2\text{O}_2$  (6 %  $\text{H}_2\text{O}_2$ )) were added to 99.43 ml DW).
- iv.  $\text{H}_2\text{SO}_4$  (5 % v/v) : 5 ml of concentrated sulphuric acid were added slowly, drop by drop, to 95 ml DW.

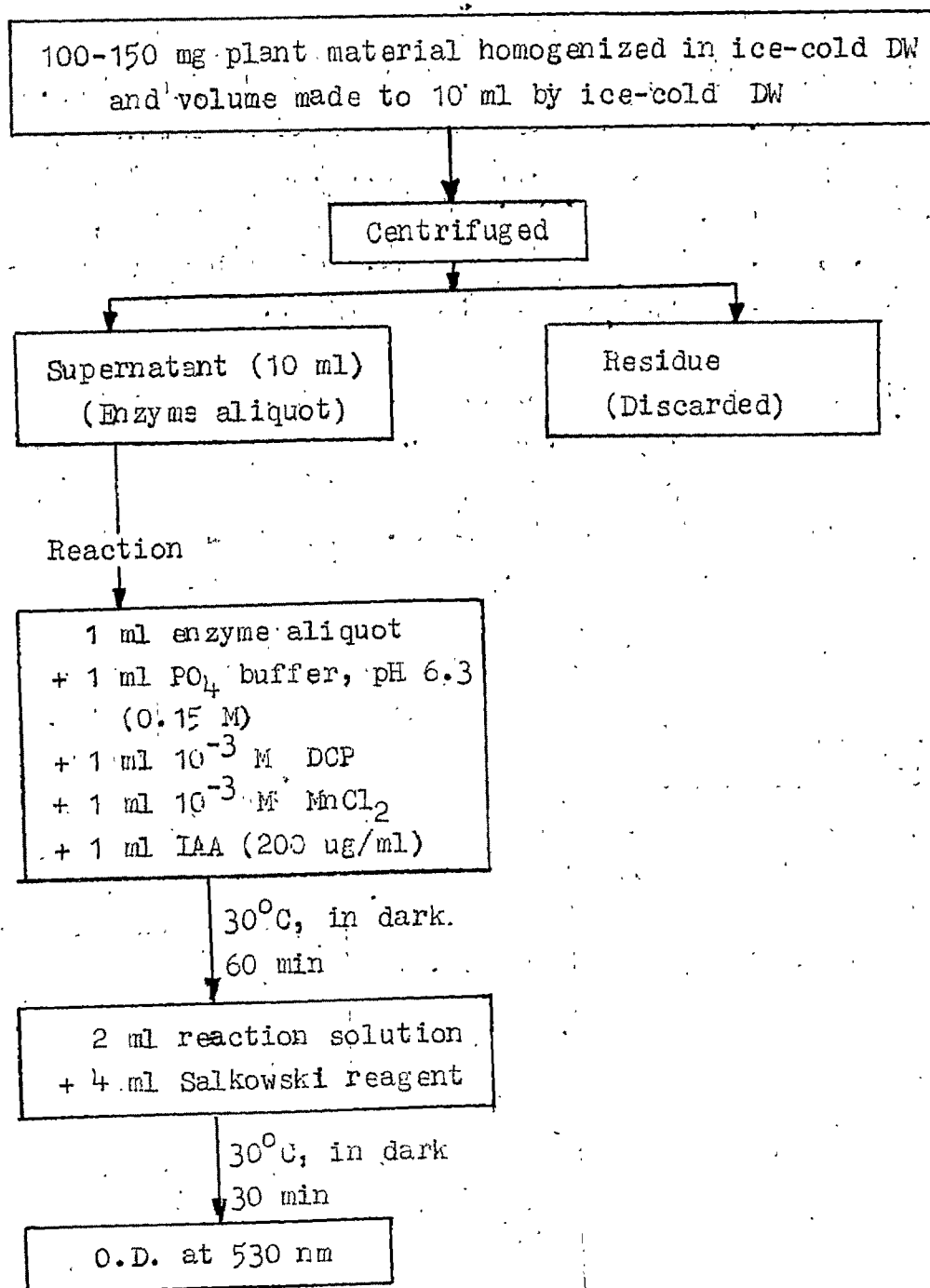
Fig. A.3 Flow Chart for Determination of Polyphenoloxidase Activity.



### Preparation of Reagents Used

- i. Phosphate buffer, pH 6.3 (0.15 M): A solution of 23.4 g  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  per litre and a solution of 21.29 g of  $\text{Na}_2\text{HPO}_4$  per litre (or 53.77 g  $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  per litre) were prepared. To make 100 ml of phosphate buffer, pH 6.3, 38.7 ml of  $\text{NaH}_2\text{PO}_4$  and 11.2 ml of  $\text{Na}_2\text{HPO}_4$  were added and volume made upto 100 ml by D.W.
- ii. Pyragallol (50  $\mu$  moles/2 ml) : 315.25 mg pyragallol was dissolved in distilled water and final volume made to 100 ml.
- iii.  $\text{H}_2\text{SO}_4$  (5 % v/v): 5 ml of concentrated sulphuric acid were added slowly drop by drop to 95 ml D.W.

Fig. A.4 Flow Chart for Determination of IAA-oxidase Activity

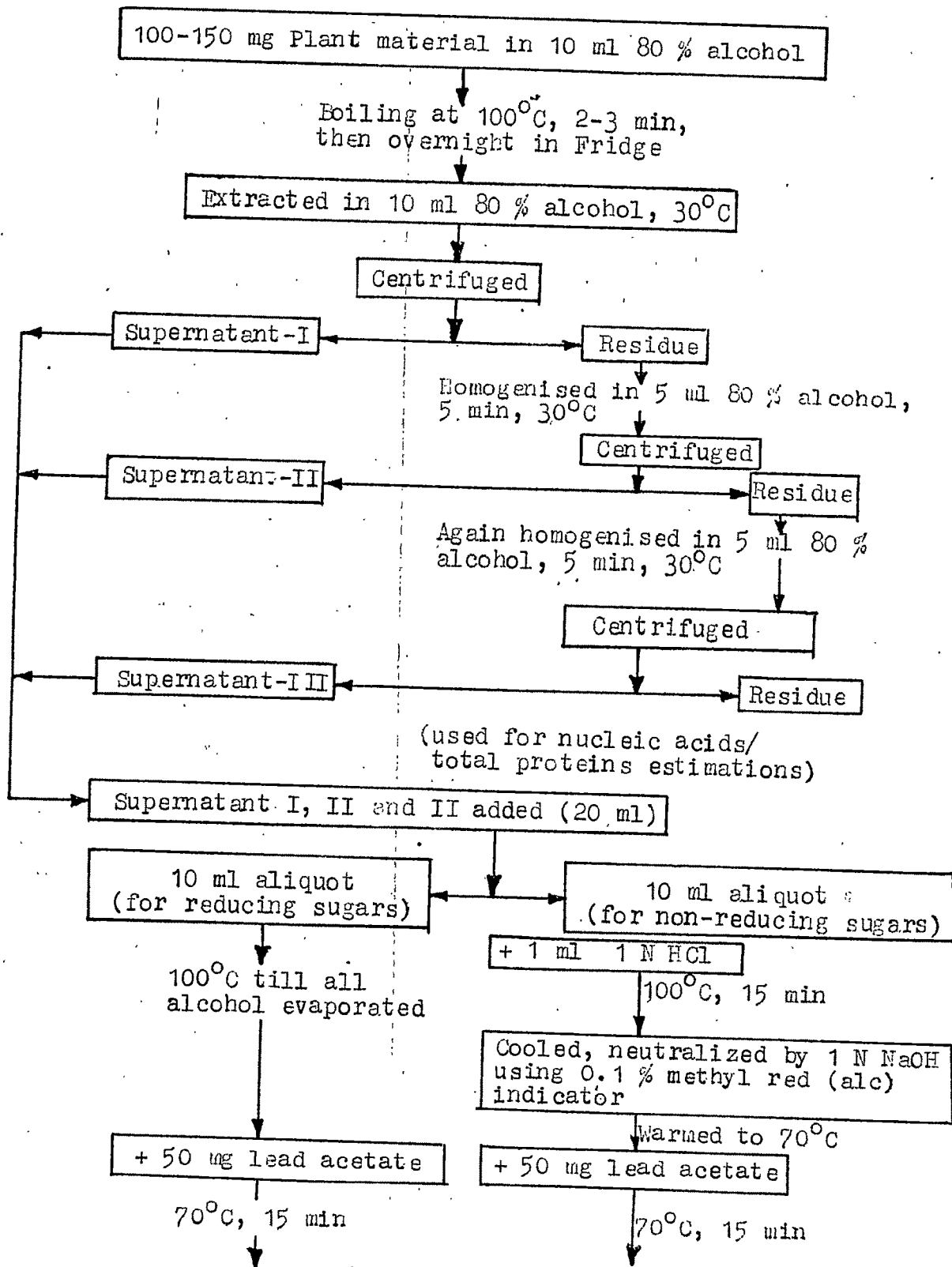


(In blank 1 ml DW was taken in place of 1 ml enzyme to estimate the O.D. due to 1 ml IAA (200 ug/ml) and in control 1 ml enzyme was taken as such but 1 ml DW was taken in place of 1 ml IAA (200 ug/ml) to minus the O.D. due to endogenous IAA if any, in enzyme extract, from the reaction OD).

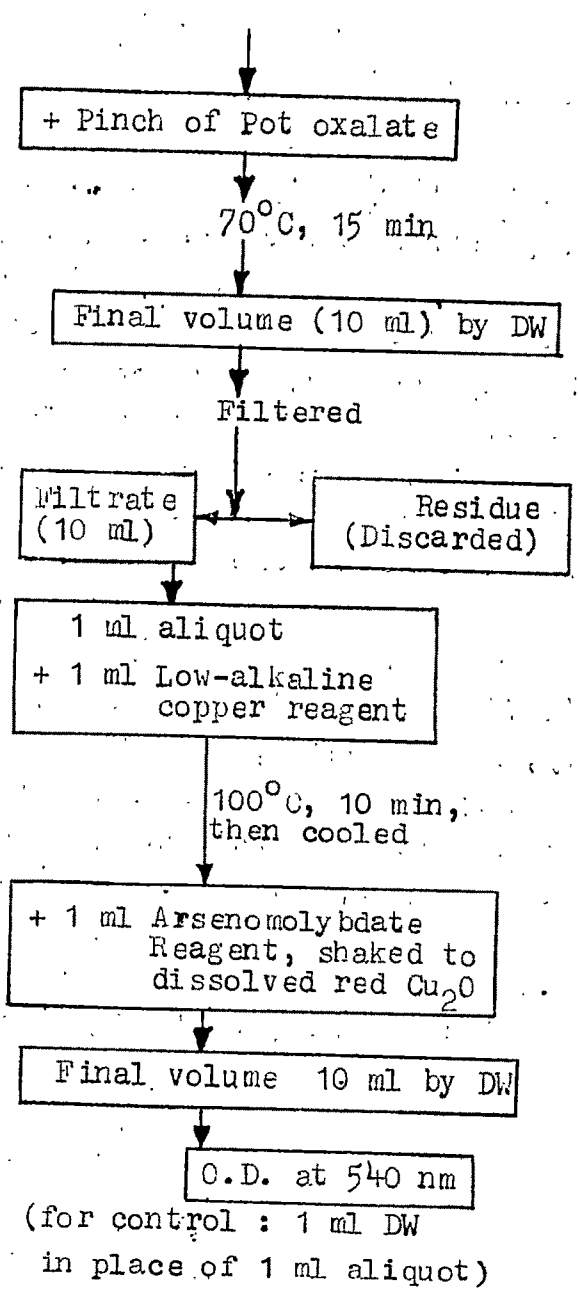
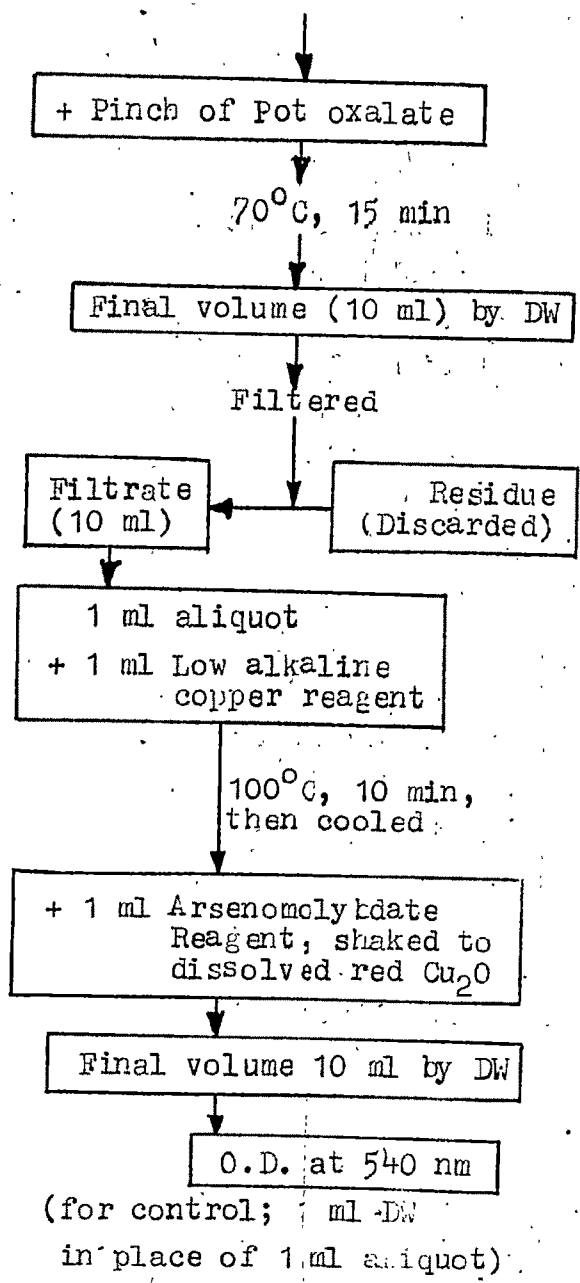
### Preparation of Reagents Used

- i. Phosphate buffer, pH 6.3 (0.15 M) : A solution of 23.4 g of  $\text{Na}_2\text{HPO}_4 \cdot 2\text{H}_2\text{O}$  per litre and a solution of 21.29 g of  $\text{Na}_2\text{HPO}_4$  per litre (or 53.77 g  $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  per litre) were prepared. To make 100 ml of phosphate buffer, pH 6.3, 38.7 ml of  $\text{NaH}_2\text{PO}_4$  and 11.2 ml  $\text{Na}_2\text{HPO}_4$  were added and volume made upto 100 ml by DW.
- ii. 2-4, Dichlorophenol (DCP) ( $10^{-3}$  M) : 16.30 mg of the DCP were dissolved in 100 ml DW.
- iii. Manganese chloride ( $10^{-3}$  M) : 19.80 mg of  $\text{MnCl}_2$  were dissolved in 100 ml of DW.
- iv. Indole-3-acetic acid (IAA) (200 ug/ml) : 20 mg of IAA was first dissolved in 5-10 drops of 95 % ethyl alcohol and then the requisite volume of DW added with vigorous stirring to the alcoholic solution, i.e, later on total volume made to 100 ml with DW.
- v. Salkowski Reagent : 2 ml of 0.5 M  $\text{FeCl}_3$  were added to 100 ml of 35 % perchloric acid.

Fig. A.5 Flow Chart for Determination of Reducing and Non-reducing Sugars



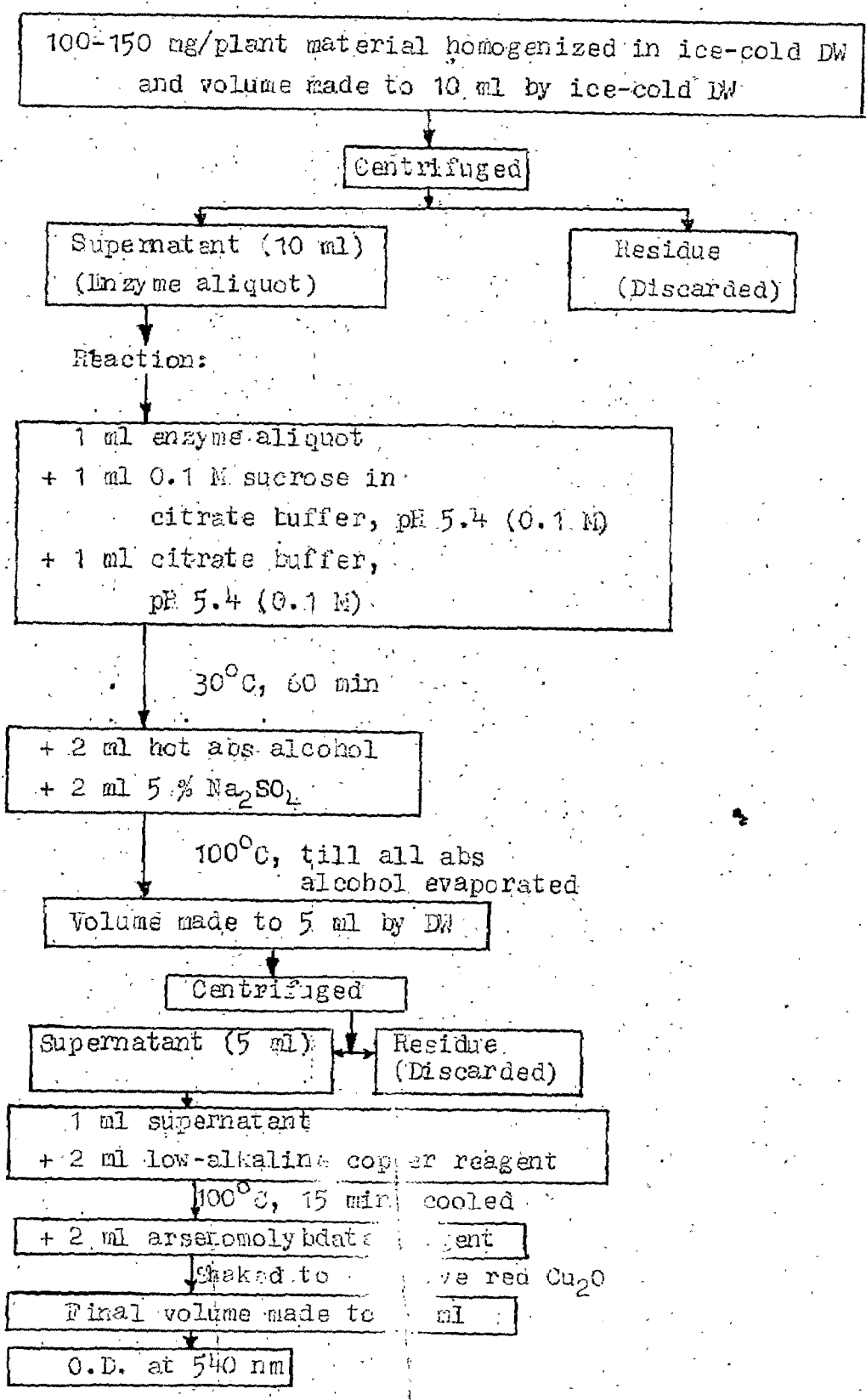




#### Preparation of Reagents Used :

- i. Low alkaline copper reagent : Rochelle salt (12 g) and anhydrous sodium carbonate (24 g) are dissolved in about 250 ml of DW. A solution of 4 g of cupric sulphate pentahydrate in DW is added with stirring, followed by 16 g of sodium hydrogen carbonate. A solution of 150 g of anhydrous sodium sulphate in 500 ml of DW is boiled to expel air, then, the two solutions are combined and diluted to 1 litre. After 1 week of standing, the clear supernatant solution is used.
- ii. Arsenomolybdate reagent : 25 g of ammonium molybdate in 450 ml of DW is added to 21 ml of conc.  $H_2SO_4$  followed by, with stirring, 3 g of disodium hydrogen arsenate heptahydrate dissolved in 25 ml of DW. The mixed solution is incubated for 24 h at  $37^\circ C$  and stored in a glass-stoppered brown bottle.
- iii. 0.1 % Methyl Red (alc) Indicator : 0.5 g of methyl red is dissolved in 300 ml of 95 % alcohol and diluted with 200 ml of DW.

Fig. A.6 Flow Chart for Determination of Invertase Activity

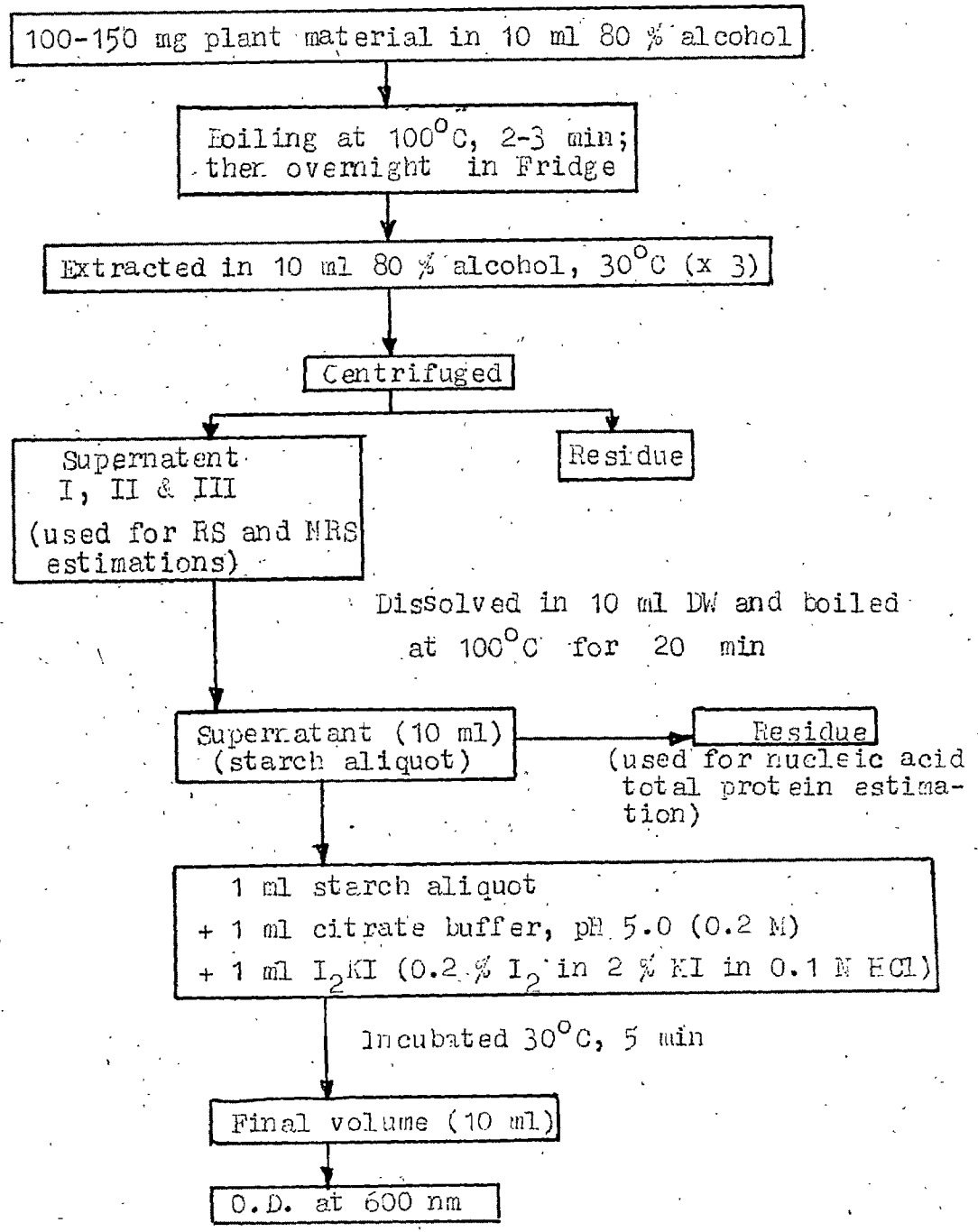


(For control, 1 ml citrate buffer, pH 5.4 was taken in place of 1 ml 0.1 M sucrose in citrate buffer, pH 5.4 to deduct the amount of reducing sugars already present in 1 ml enzyme extract)

#### Preparation of reagents Used

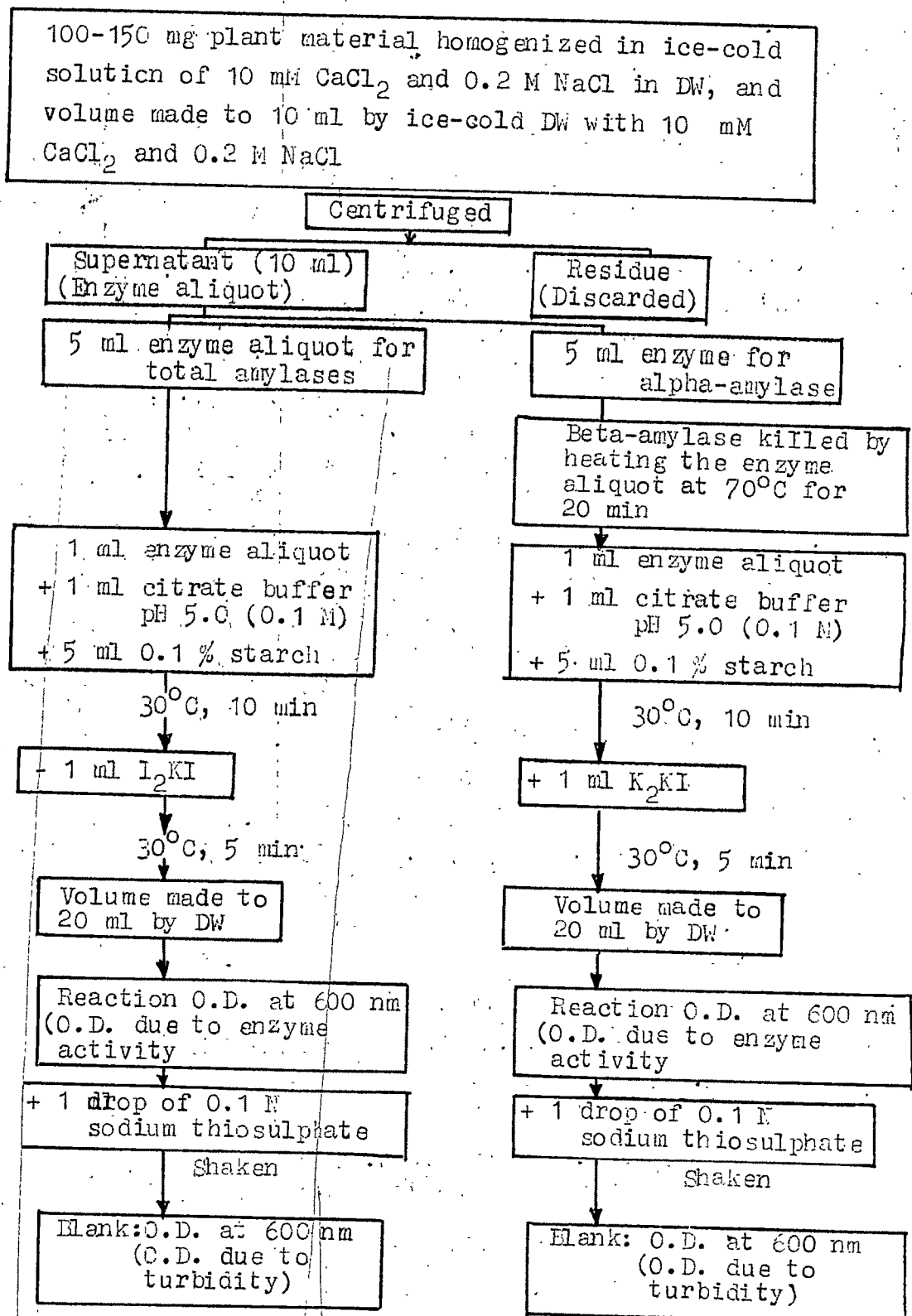
- i. Citrate buffer, pH 5.4 (0.1 M) : A 0.1 M solution of citric acid (19.21 g per litre) and a 0.1 M solution of sodium citrate (29.41 g/litre) were prepared. To make 100 ml citrate buffer, pH 5.4, 16.0 ml of citric acid and 34.0 ml sodium citrate were added and volume made upto 100 ml DW.
- ii. 0.1 M Sucrose in citrate buffer, pH 5.4 (0.1 M) : 3.42 g of sucrose were dissolved in 90 ml of citrate buffer, pH 5.4 (0.1 M) and total volume made to 100 ml by citrate buffer, pH 5.4 (0.1 M).
- iii. 5 % sodium sulphate : 5 g sodium sulphate dissolved in 100 ml DW.
- iv. Low-alkaline copper reagent : as described earlier in methods of reducing and non-reducing sugars (Fig. ).
- v. Arsenomolybdate reagent : as described earlier in methods of reducing and non-reducing sugars (Fig. ).

Fig. A.7 Flow Chart for Determination of Starch



(for control : 1 ml DW is taken in place of 1 ml starch aliquot)

Fig. A.8 Flow Chart for Determination of Alpha and Beta Amylase Activities

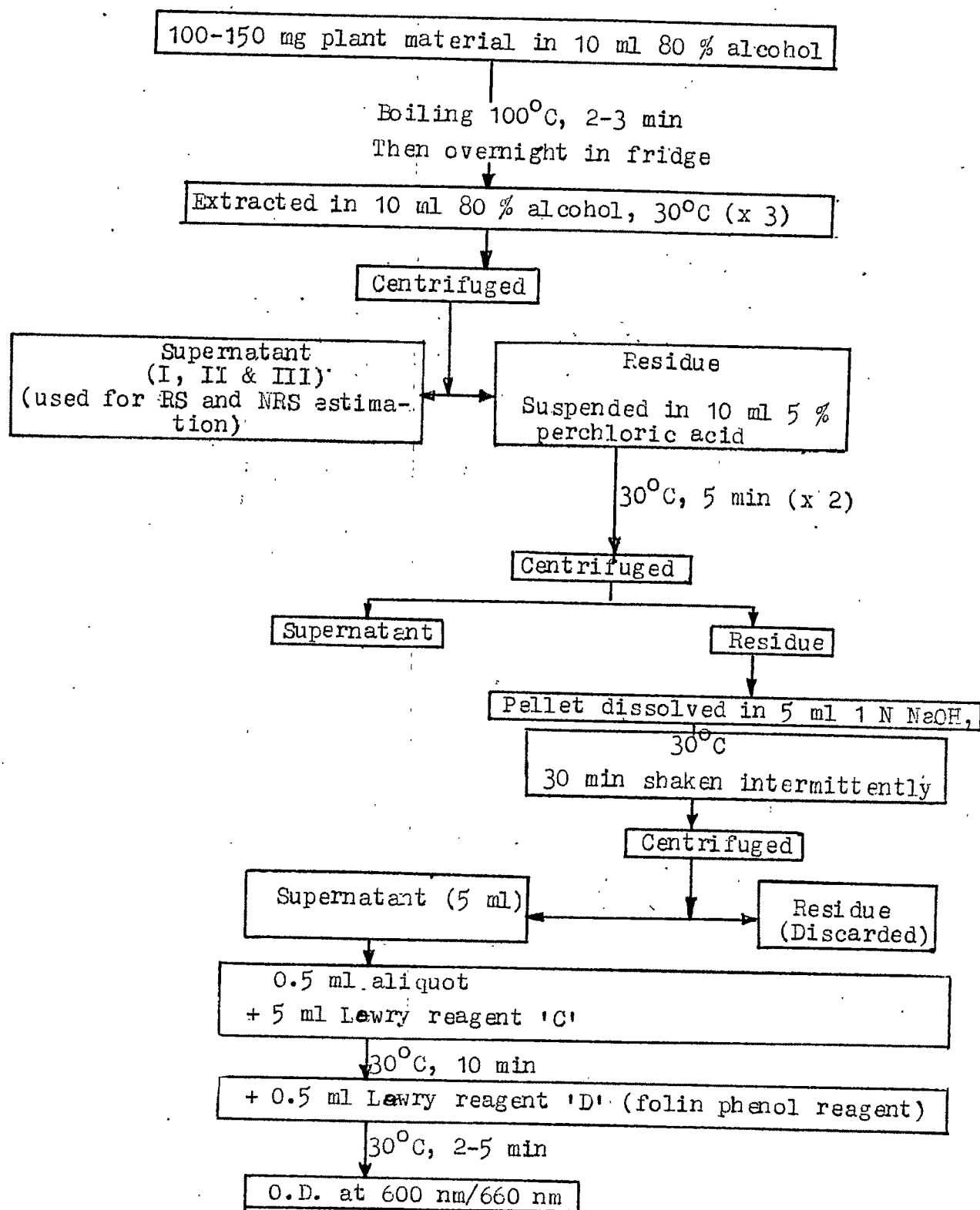


Control was prepared in the same way with 1 ml 10 mM  $\text{CaCl}_2$  in 0.2 M NaCl solution in DW instead of 1 ml enzyme)

#### Preparation of Reagents Used

- i. Citrate buffer, pH 5.0 (0.1 M) : Made up a 0.1 M solution of citric acid (19.21 g per litre) and a 0.1 M solution of sodium citrate (29.41 g per litre). To make 100 ml citrate buffer, pH 5.0, 20.5 ml of citric acid and 29.5 ml sodium citrate were added and volume made upto 100 ml with DW.
- ii. 0.1 % starch : 100 mg of starch powder boiled for 2-3 minutes in 90 ml DW till the solution becomes 100 % transparent, cooled and volume made upto 100 ml by DW.
- iii.  $\text{I}_2$  KI reagent (0.2 %  $\text{I}_2$  in 2 % KI in 0.1 N HCl) : 200 mg fresh iodine crystals dissolved in 2 % potassium iodide solution prepared in 0.1 N HCl (for 100 ml) of 0.1 N HCl, 0.88 ml conc HCl (35 % HCl) were dissolved in DW to make upto 100 ml volume).
- iv. 0.1 N sodium thiosulphate : 2.48 g of sodium thiosulphate (MW - 248.18) were dissolved in 100 ml DW.
- v. 10 mM  $\text{CaCl}_2$  in 0.2 M NaCl : 11.69 NaCl and 1.11 g  $\text{CaCl}_2$  were dissolved in DW and volume made to 1 litre.

Fig. A.9 Flow Chart for Determination of Total Proteins



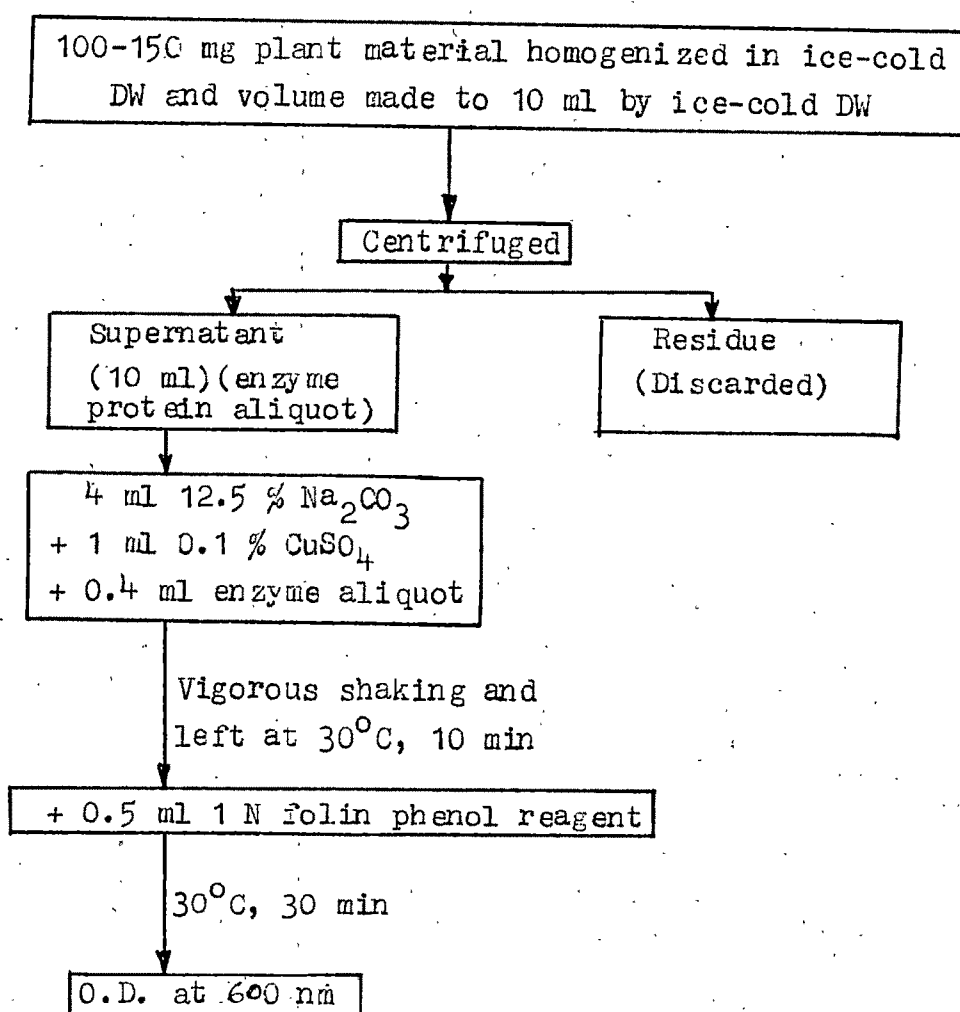
(For control : 0.5 ml 1 N NaOH is taken in place of 0.5 ml protein aliquot)



### Preparation of Reagents Used

- i. Lewry Reagent 'C' : Lewry Reagent 'A' (2 %  $\text{Na}_2\text{CO}_3$  in 0.1 N NaOH) was prepared by adding 2 g  $\text{Na}_2\text{CO}_3$  in 0.1 N NaOH and made to a volume 100 ml. (0.1 N NaOH was prepared by adding 0.4 g NaOH in 100 ml DW). Lewry Reagent 'B' (0.5 %  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in 1 % sod. or pot. tartrate) was prepared by dissolving 500 mg  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  and 1 g sod. or pot. tartrate in distilled water and made the volume to 100 ml. For Lewry Reagent 'C' 50 parts of Lewry Reagent 'A' and 1 part of Lewry Reagent 'B' were added.
- ii. Lewry Reagent 'D' : 1 part of 3 N foline phenol Reagent Folin Ciocalteus Reagent = Phenol-indo-2,6-dichlorophenol) were added with 2 part of distilled water.
- iii. 5 % Perchloric acid : 5 ml of 70 % perchloric acid were added to distilled water and final volume made to 70 ml by DW.

Fig. A.10 Flow Chart for Determination of Enzyme Proteins

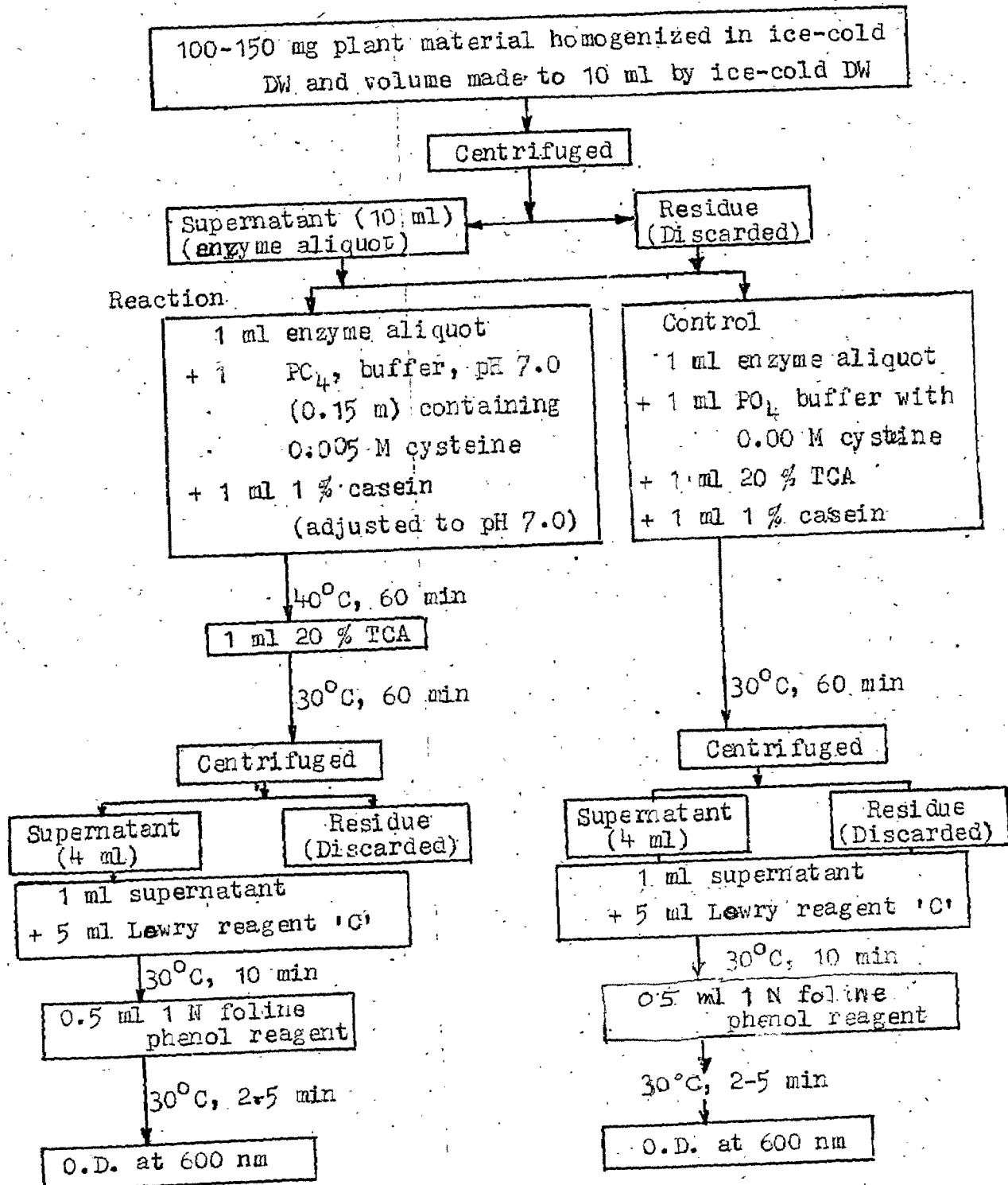


(For Blank : 0.4 ml DW was taken in place 0.4 ml enzyme protein aliquot)

#### Preparation of Reagents Used

- i. 12.5 %  $\text{Na}_2\text{CO}_3$  : 12.5 sodium carbonate dissolved in 100 ml DW.
- ii. 0.1 %  $\text{CuSO}_4$  : 100 mg of copper sulphate were dissolved in 100 ml DW.
- iii. 1 N Folin phenol reagent : 1 part of folin phenol reagent was added to 2 parts of DW.

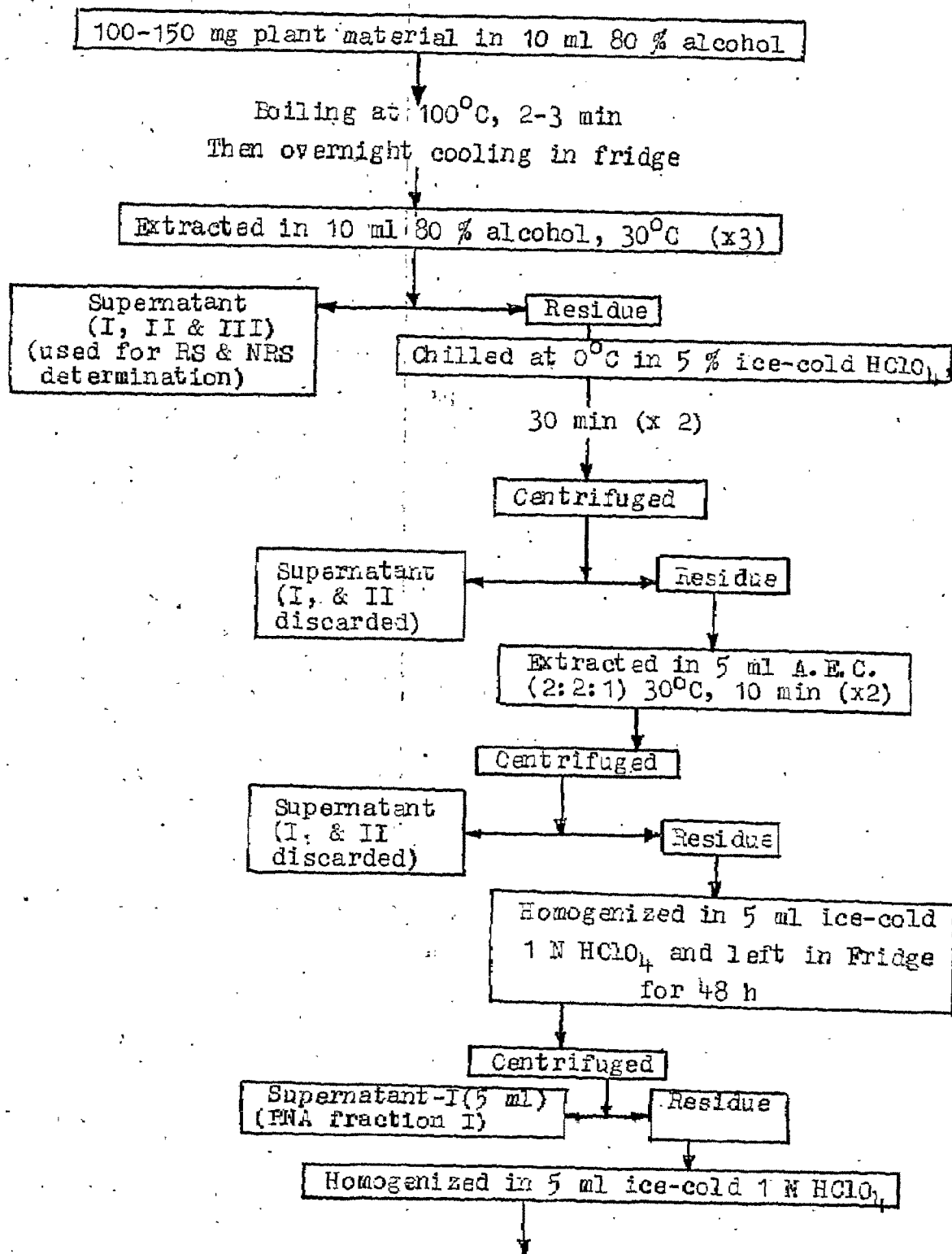
Fig. A.11 Flow chart for Determination of Protease Activity



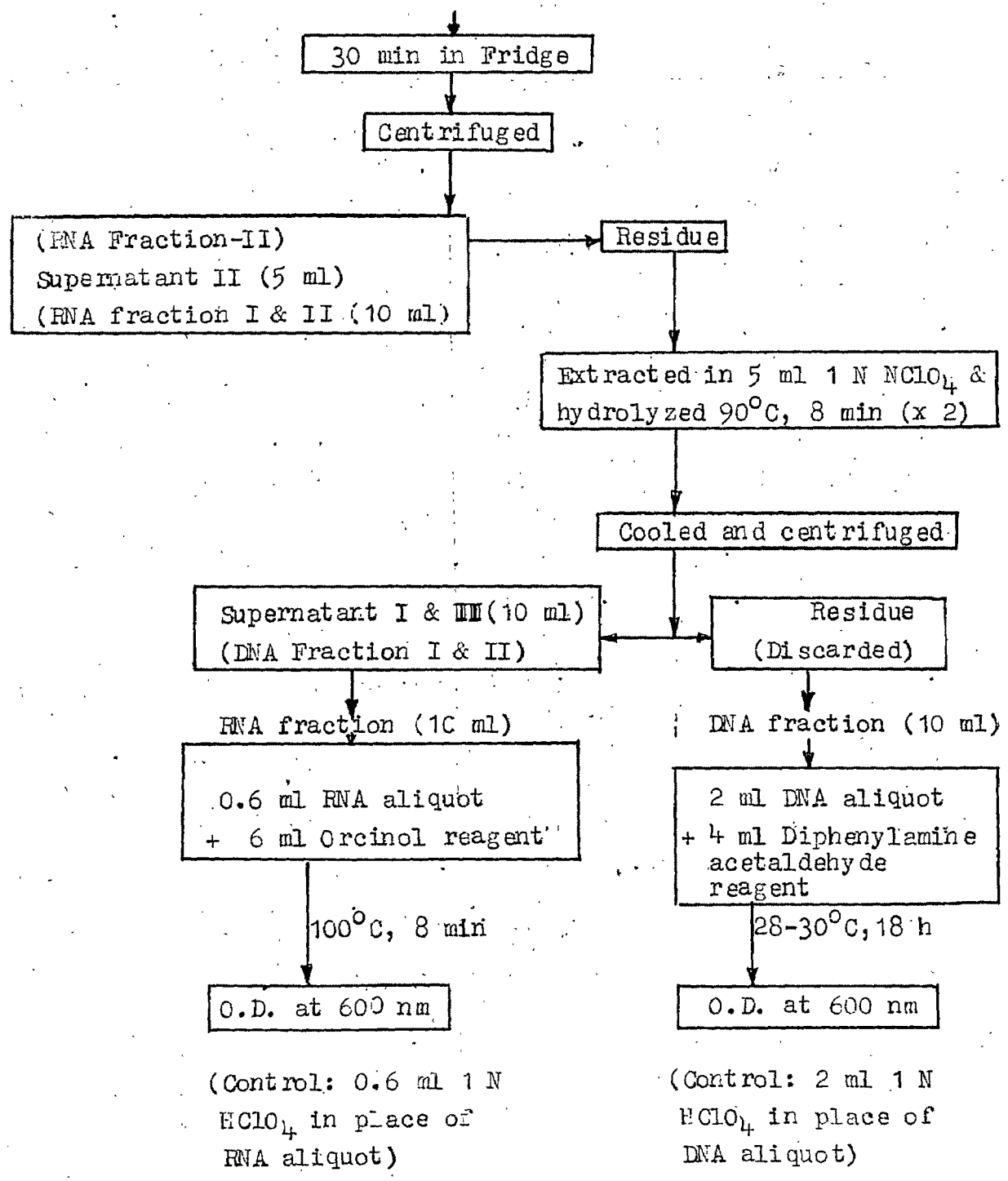
#### Preparation of Reagents Used

- i. Phosphate buffer, pH 7.0 (0.15 M) containing 0.005 M cysteine: A solution of 23.4 g of  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  per litre and a solution of 21.29 g of  $\text{Na}_2\text{HPO}_4$  per litre (or 53.77 g  $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  per litre) were prepared. To make 100 ml of phosphate buffer, pH 7.0, 19.5 ml of  $\text{NaH}_2\text{PO}_4$  and 30.5 ml of  $\text{Na}_2\text{HPO}_4$  were added and volume made up to 100 ml by DW. To prepare 0.005 M cysteine (MW 121.16) solution, 60.50 mg -cysteine/100 ml was added to the phosphate buffer or for 0.005 M -cysteine hydrochloride monohydrate (M.W. 175.64); 97.82 mg was added to 100 ml  $\text{PO}_4$  buffer.
- ii. 20 % TCA : 20 g of trichloroacetic acid dissolved in 100 ml DW.
- iii. Lowry reagent A: 2 %  $\text{Na}_2\text{CO}_3$  in 0.1 N NaOH Lowry reagent 'B' : 0.5 %  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in 1 % Na or K-tartrate. Fresh Lowry reagent 'C' was prepared by adding 50 ml of 'A' and 1 ml of 'B'.
- iv. 1 N folin phenol reagent (Lowry reagent 'D') : 1 part of folin phenol reagent + 2 parts of DW.
- v. 1 % Casein : 1 gm casein first dissolved in 2-3 ml 0.1 N NaOH, then diluted to 100 ml volume by phosphate buffer, pH 7.0.

Fig. A.12 Flow Chart for Determination of Nucleic Acid (RNA and DNA)



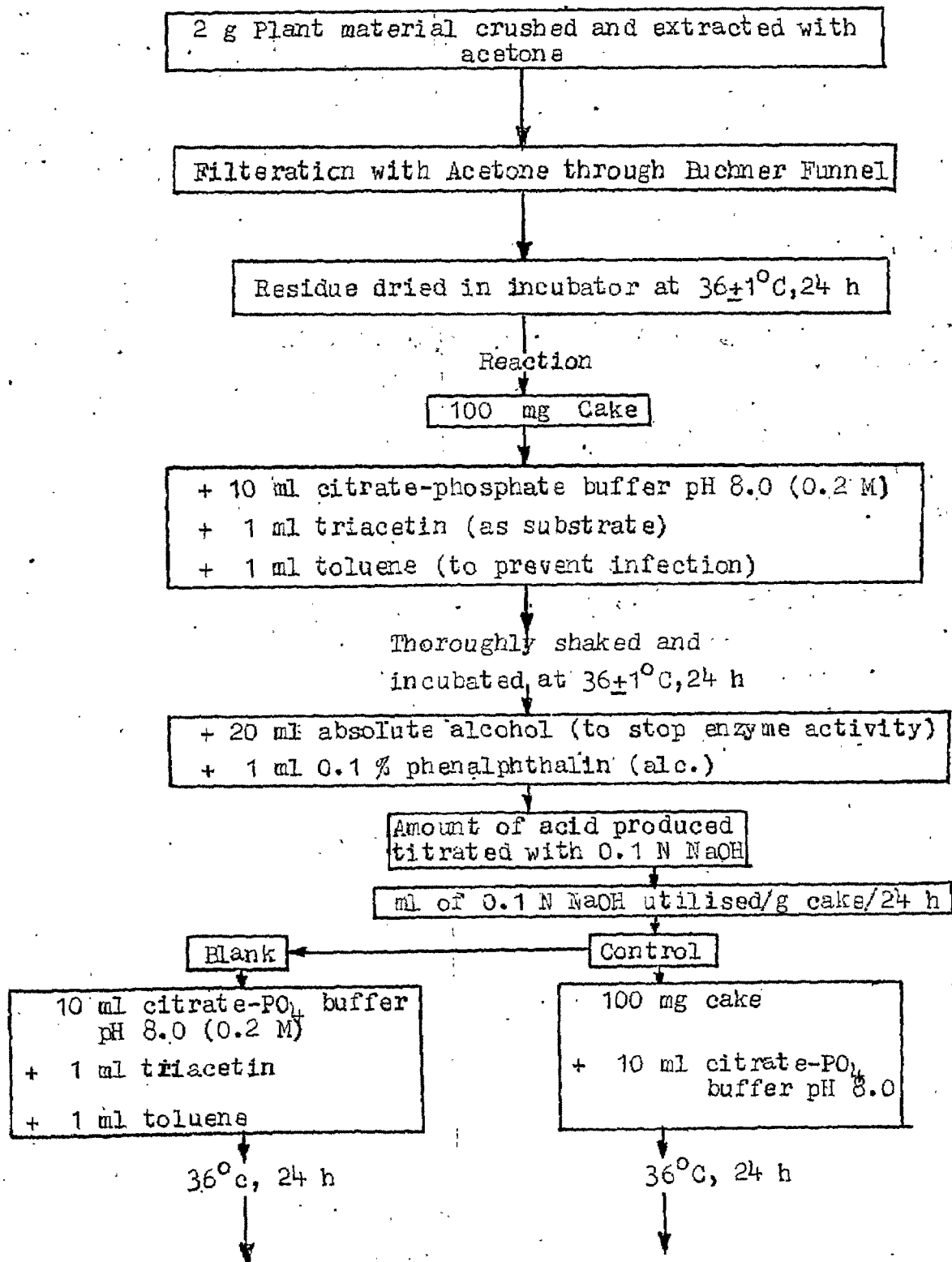
CONTD.



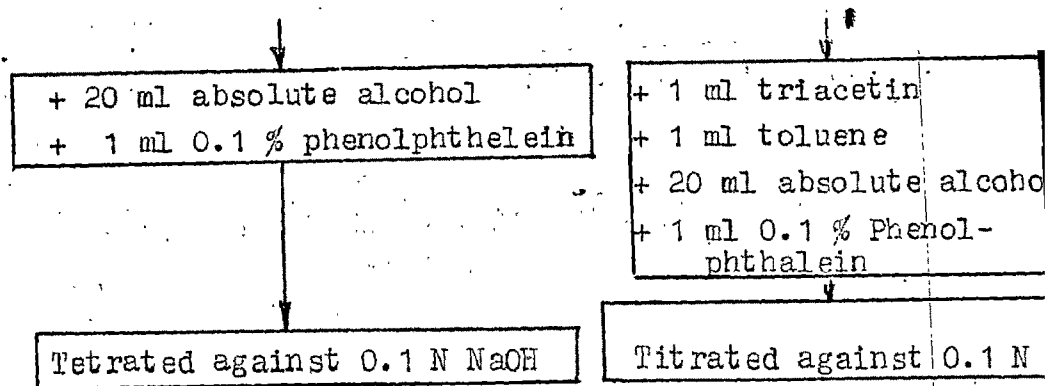
#### Preparation of Reagents Used

- i. Orcinol reagent (after Markham, 1955) : The fresh orcinol reagent was prepared by mixing 10 volume of 1 % orcinol with 40 ml volume of con. HCl and 1 volume of 10 %  $\text{FeCl}_3 \cdot 3\text{H}_2\text{O}$ .
- ii. Diphenylamine acetaldehyde reagent (after Burton, 1956) : Diphenylamine reagent was prepared by dissolving 6 g diphenylamine in 460 ml acetic acid followed by addition of 6 ml conc  $\text{H}_2\text{SO}_4$ . 0.1 ml of 16 mg/ml acetaldehyde was added to each 20 ml of the diphenylamine reagent before use (16 mg/ml acetaldehyde is prepared by adding 0.2 ml pure acetaldehyde in 9.8 ml DW).
- iii. 5 % Perchloric acid : 7.14 ml of 70 % perchloric acid is added with 92.9 ml DW (or 5 ml of 70 % perchloric acid taken and volume made to 70 ml by DW).
- iv. 1 N perchloric acid : 11.1 ml of 60 % perchloric acid added with 89.9 ml DW or 9.0 ml of 70 % perchloric acid added with 91.0 ml DW.
- v. Alcohol : Ether : Chloroform in 2:2:1 ratio.

Fig. A.13 Flow Chart for Determination of Lipase Activity



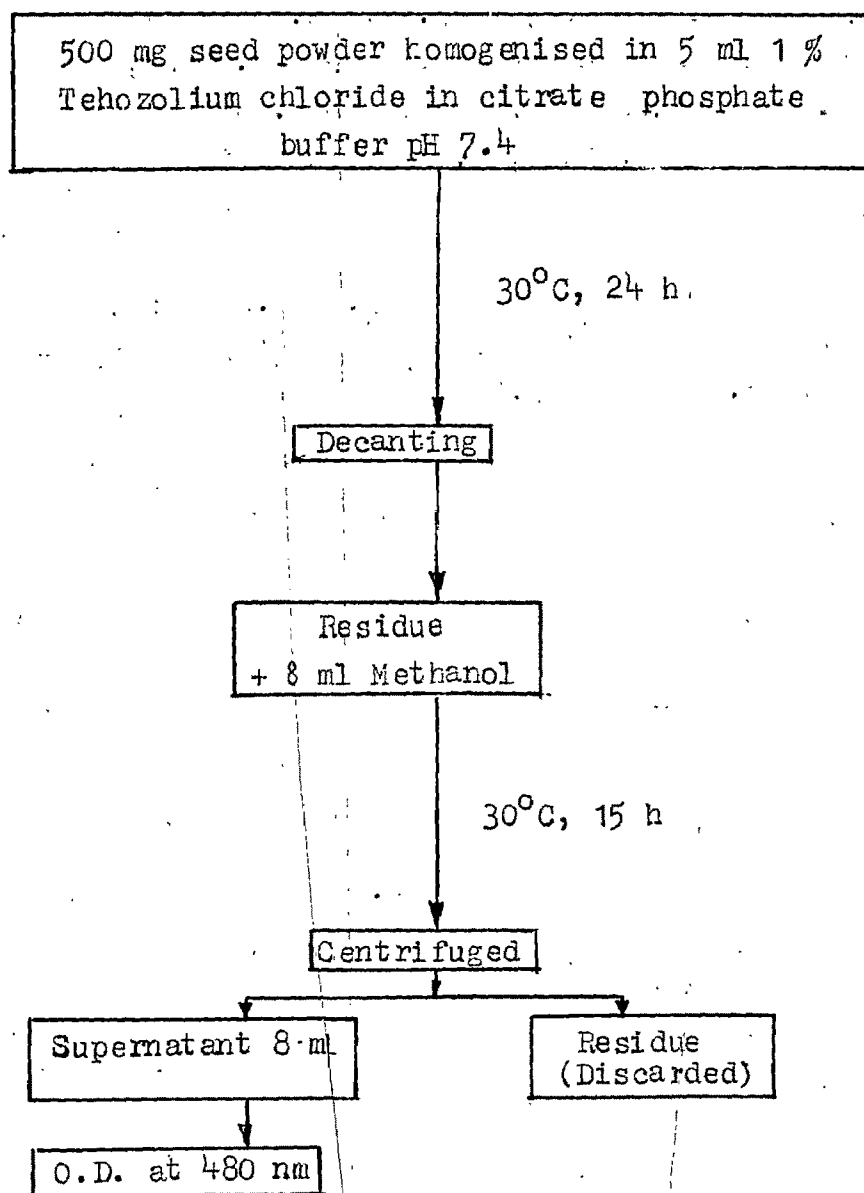




#### Preparation of Reagents Used

- i. 0.1 N NaOH : 4 g NaOH (Mol wt. 40) dissolved in 1 litre DW.
- ii. 0.1 % phenolphthalein (alc.) : 100 mg of phenolphthalein dissolved in 100 ml 95 % alcohol.
- iii. Citrate-phosphate buffer, pH 8.0 (0.2 M) ; 97.25 ml of disodium hydrogen orthophosphate (Mol wt. 142), 0.2 M (2.83 g in 100 ml DW) added to 2.75 ml of citric acid (Mol. wt. 210), 0.1 M (2.10 g in 100 ml DW).

Fig. A.14 Flow Chart for Determination of Dehydrogenase Activity

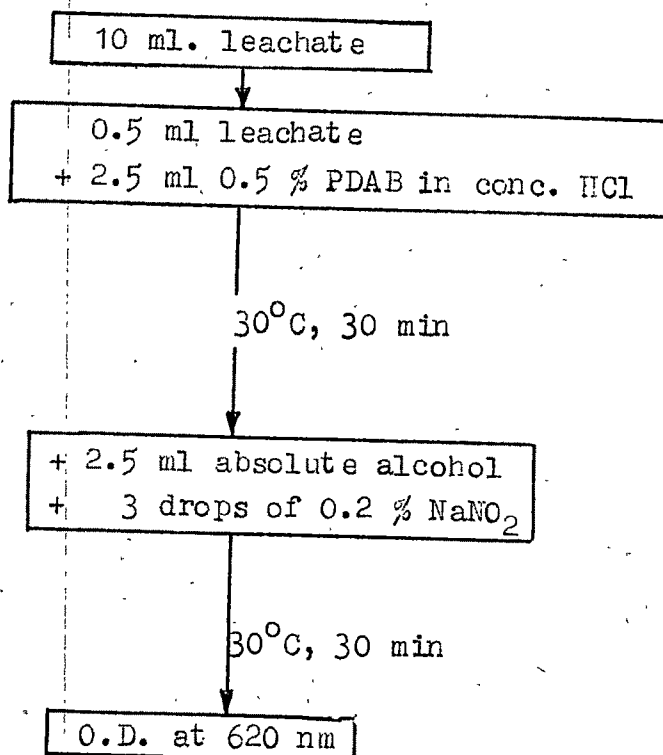


(For Blank : Pure 8 ml methanol was taken)

Preparation of Reagents Used

- i. Citrate phosphate buffer, pH 7.4 : 91.85 ml of disodium hydrogen orthophosphate (Mol wt. 142), 0.2 M (2.83 g in 100 ml DW) added to 9.15 ml of citric acid (Mol wt. 210), 0.1 M (2.10 g in 100 ml DW).
- ii. 1 % TTC in citrate- $\text{PO}_4$  Buffer pH 7.4 (0.2 M) : 1 g Tetrazolium chloride dissolved in 100 ml of citrate- $\text{PO}_4$  buffer (0.2 M) pH 7.4

Fig. A.15 Flow Chart for Determination of Tryptophan



## Preparation of Reagents Used

- i. 0.5 % PDAB (Para-di-amino-benzaldehyde) in conc HCl : 50 mg of dye were dissolved in 10 ml conc HCl.
- ii. 0.2 % sodium nitrite : 20 mg of sodium nitrite were dissolved in 10 ml DW.