

Abstract of the Thesis entitled

“Structural, Magnetic, Catalytic And Spectroscopic Properties Of Metal Complexes Of Varying Nuclearity”

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This thesis includes the contents of seven published articles: *Cryst. Growth. Des.* **2016**, *16*, 3777–3790; *Polyhedron* **2015**, *87*, 98–108, **2014**, *77*, 39–46, and **2014**, *73*, 67–71; *Inorg. Chim. Acta* **2017**, *455*, 70–80, **2015**, *435*, 38–45, and **2014** *410*, 65–75.

This thesis is comprised of eight chapters. Brief reviews of the relevant topics and objective and scope of the thesis are discussed in **Chapter 1**.

Eight single / double compartmental Schiff base ligands have been used. Each ligand is a 2:1 aldehyde:diamine condensation product: 3-methoxysalicylaldehyde–ethylenediamine (H_2L^1), 3-ethoxysalicylaldehyde–ethylenediamine (H_2L^2), 3-methoxysalicylaldehyde–*o*-phenylenediamine (H_2L^3), 3-methoxysalicylaldehyde–*trans*-1,2-diaminocyclohexane (H_2L^4), 2-hydroxyacetophenone–2,2-dimethyl-1,3-diaminopropane (H_2L^5), salicylaldehyde–2,2-dimethyl-1,3-diaminopropane (H_2L^6), 3-ethoxysalicylaldehyde–2,2-dimethyl-1,3-diaminopropane (H_2L^7), and salicylaldehyde–1,3-diaminopropane (H_2L^8). All the thirty five (35) compounds that are reported in this thesis in **Chapters 2–8** have been synthesized and characterized by single crystal X-ray structure determination. In addition to structural studies, other studies/aspects include: Catechol oxidase activity (**Chapters 4 and 5**); ESI-MS (**Chapters 4, 5, and 7**), magnetic properties (**Chapter 8**), lone pair functionality (**Chapter 7**), electrochemistry (**Chapter 4**), and diffuse reflectance spectra (**Chapters 3 and 6**).

Chapter 2 deals with ten $Cu^{II}Sn^{IV}$ / $Cu^{II}Sn^{II}$ salts / cocrystals / salt cocrystals of composition $[CuL^1]_2 \cdot [SnMe_2Cl_4]^{2-} \cdot (H_2ED)^{2+}$ (**1**), $[CuL^1]_2 \cdot [SnEt_2Cl_4]^{2-} \cdot (H_2ED)^{2+} \cdot 0.5H_2O$ (**2**), $[CuL^1]_2 \cdot [Sn(n-Bu)_2Cl_4]^{2-} \cdot (H_2ED)^{2+}$ (**3**), $[CuL^1]_2 \cdot [SnPh_2Cl_4]^{2-} \cdot (H_2ED)^{2+}$ (**4**), $[CuL^1]_2 \cdot [SnPh_2Cl_4]^{2-} \cdot (H_2ED)^{2+} \cdot 2MeOH$ (**5**), $[CuL^2]_2 \cdot [SnMe_2Cl_2(H_2O)_2] \cdot 0.2H_2O$ (**6**), $[CuL^2] \cdot [Sn(n-Bu)_2Cl_2(H_2O)]$ (**7**), $[CuL^2]_2 \cdot [SnPh_2Cl_4]^{2-} \cdot (H_2ED)^{2+}$ (**8**), $[CuL^1SnCl]^+ \cdot [SnCl_3]^-$ (**9**), and $[CuL^2SnCl]^+ \cdot [SnCl_3]^-$ (**10**) (ED = 1,2-ethylenediamine).

Chapter 3 deals with $\{[Cu^{II}L^1](Me_2NH_2)\}(ClO_4)$ (**11**); $\{[Cu^{II}L^3](Me_2NH_2)\}(ClO_4)$ (**12**); $\{[Cu^{II}L^4](Me_2NH_2)\}(ClO_4)$ (**13**); $\{[Cu^{II}L^1]_2(enH_2)\}(ClO_4)_2$ (**14**); $\{[Cu^{II}L^3]_2(opdaH_2)\}(ClO_4)_2 \cdot 2H_2O$ (**15**) (enH_2 / $opdaH_2$ = diprotonated ethylenediamine / *o*-phenylenediamine); $\{[Cu^{II}L^4](MeOH_2)\}(ClO_4)$ (**16**).

Chapter 4 deals with $[\text{Mn}^{\text{III}}\text{L}^2(\text{OAc})(\text{H}_2\text{O})]\cdot\text{DMF}$ (**17**), $[\text{Mn}^{\text{III}}\text{L}^2(\text{Cl})(\text{H}_2\text{O})]\cdot 2\text{H}_2\text{O}$ (**18**), $[\text{Mn}^{\text{III}}\text{L}^2(\text{N}_3)(\text{H}_2\text{O})]$ (**19**), $[\text{Mn}^{\text{III}}\text{L}^2(\text{NCS})(\text{H}_2\text{O})]$ (**20**), and $[\text{Mn}^{\text{III}}\text{L}^2(\text{NCSe})(\text{H}_2\text{O})]\cdot\text{MeCN}$ (**21**).

Chapter 5 deals with $[\text{Co}^{\text{III}}\text{L}^2(\text{N}_3)_2\supset(\text{H}_3\text{O}^+)]\cdot 2\text{MeOH}$ (**22**) and $[\text{Co}^{\text{III}}\text{L}^2(\text{NCS})(\text{H}_2\text{O})]\cdot\text{DMF}\cdot\text{H}_2\text{O}$ (**23**).

Chapter 6 deals with of two mononuclear compounds of composition $[\text{Cu}^{\text{II}}\text{L}^5]\cdot\text{MeOH}$ (**24**) and $[\text{Ni}^{\text{II}}\text{L}^5]$ and six heteronuclear complexes of composition $[\text{Cu}^{\text{II}}(\text{acetone})\text{L}^5(\text{U}^{\text{VI}}\text{O}_2)(\text{NO}_3)_2]\cdot 1.5\text{CH}_3\text{COCH}_3$ (**25**), $[\text{Cu}^{\text{II}}(\text{H}_2\text{O})\text{L}^7(\text{U}^{\text{VI}}\text{O}_2)(\text{NO}_3)_2]\cdot\text{MeCN}$ (**26**), $[\text{Ni}^{\text{II}}\text{L}^5(\text{U}^{\text{VI}}\text{O}_2)(\text{NO}_3)_2]\cdot\text{CH}_3\text{COCH}_3$ (**27**), $[\{\text{Cu}^{\text{II}}\text{L}^5\text{Ag}^{\text{I}}(\text{NO}_3)\}_2]$ (**28**), $[\{\text{Cu}^{\text{II}}\text{L}^6\text{Ag}^{\text{I}}(\text{NO}_3)\}_2]$ (**29**), and $[(\text{Cu}^{\text{II}}\text{L}^6)_2\text{Cd}^{\text{II}}(\text{ClO}_4)_2]$ (**30**).

Chapter 7 deals with of two trinuclear compounds, $[(\text{Cu}^{\text{II}}\text{L}^8)_2\text{Pb}^{\text{II}}(\text{ClO}_4)_2]$ (**31**) and $[(\text{Cu}^{\text{II}}\text{L}^6)_2\text{Pb}^{\text{II}}(\text{NO}_3)_2]$ (**32**), one dimer of trinuclear compound, $[\{(\text{Cu}^{\text{II}}\text{L}^6)_2\text{Pb}^{\text{II}}\}_2(\mu\text{-adipate})](\text{ClO}_4)_2\cdot 2\text{H}_2\text{O}$ (**33**), and one trinuclear based 1-D coordination polymer, $[(\text{Cu}^{\text{II}}\text{L}^6)_2\text{Pb}^{\text{II}}(\mu_{1,5}\text{-dicyanamide})_2]_n$ (**34**).

Chapter 8 deals with one-dimensional coordination polymer of copper(II) of composition $[\text{Cu}^{\text{II}}_6(\text{N,N}\text{-diEten})_2(\mu_{1,1}\text{-N}_3)_8(\mu_{1,1,1}\text{-N}_3)_2(\mu_{1,3}\text{-N}_3)_2]_n$ (**35**), where N,N-diEten is N,N-diethylethylenediamine.

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