a) Most of the reactions were carried out in oven dried glasswares.
b) The common reagents were purchased from Sigma–Aldrich, Lobachemie and CDH, SD fine, India and used without further purification.
c) All the evaporations were carried out under reduced pressure on Equitron rota evaporator.
d) The progress of the reactions was checked by thin layer chromatography (TLC) using Merck silica gel (60 F$_{254}$) on aluminium plates.
e) Nuclear magnetic resonance spectroscopy was performed using Varian EM-360L and Bruker DPX 200, 300, 400 or 500 MHz spectrometer using TMS as an internal standard.
f) Chemical shifts have been expressed in δ values downfield from TMS. Multiplicity of NMR signals is designated as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), b (broad). $^{13}$C NMR Spectra were recorded with complete proton decoupling.
g) Mass spectra were recorded on ESI-esquire 3000 Bruker Daltonics instrument.
h) IR spectra (KBr discs) were recorded on a THERMO FT-IR (Nicolet-380) spectrometer.
i) UV spectra were recorded on SHIMADZU UV-1700 (double-beam) spectrophotometer.
j) Melting points were obtained on Decibel digital melting point apparatus in open glass capillary tubes and were uncorrected.